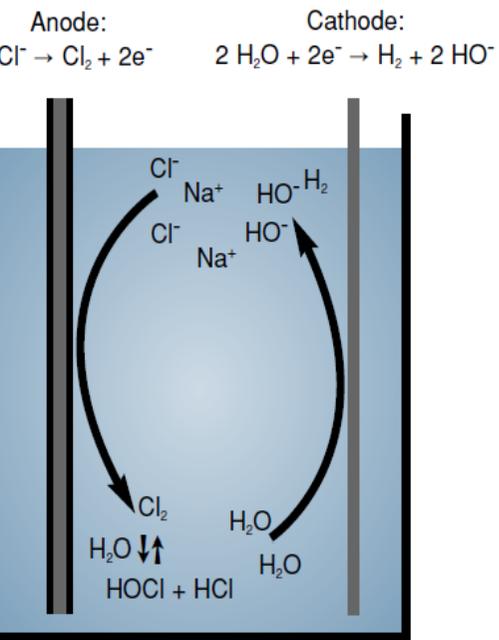


Development of HOCl generator



Dissolution of NaCl	$\text{NaCl} \rightarrow \text{Na}^+ + \text{Cl}^-$
Anode Reactions	<u>Electrochemical process</u>
	$2\text{H}_2\text{O} \rightarrow \text{O}_2 + 4\text{H}^+ + 2\text{e}^-$, $E^0 = 1.23\text{ V}$
	$2\text{Cl}^- \rightarrow \text{Cl}_2 + 2\text{e}^-$, $E^0 = 1.36\text{ V}$
	<u>Chemical process</u>
	$\text{Cl}_2 + \text{H}_2\text{O} \rightarrow \text{HCl} + \text{HOCl}$
Cathode Reactions	$2\text{Na}^+ + 2\text{H}_2\text{O} + 2\text{e}^- \rightarrow \text{H}_2 + 2\text{NaOH}$, $E^0 = -0.83\text{ V}$



Photograph of the unit developed jointly by BTDG and CG

Schematic representation of the electrochemical processes at anode and cathode

1. The EW is produced from electrolysis of 0.1% NaCl solution.
2. The HOCl content in EW is 100-200 ppm and pH range 5.5 to 6.5.
3. The anti-bacterial action of above mentioned EW has been tested by FTD, BSG and instantaneous anti-bacterial action has been observed.